	Shri Manibhai Virani and Smt. Navalben Virani Science College An Autonomous college affiliated to Saurashtra University, Rajkot	NAAC – Cycle - 3
		Criterion- IV
		Metric - 4.2.2

4.2.2	E-resources Subscription (Sister Institute)
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Figure 1: IEEE Xplore online resource home page

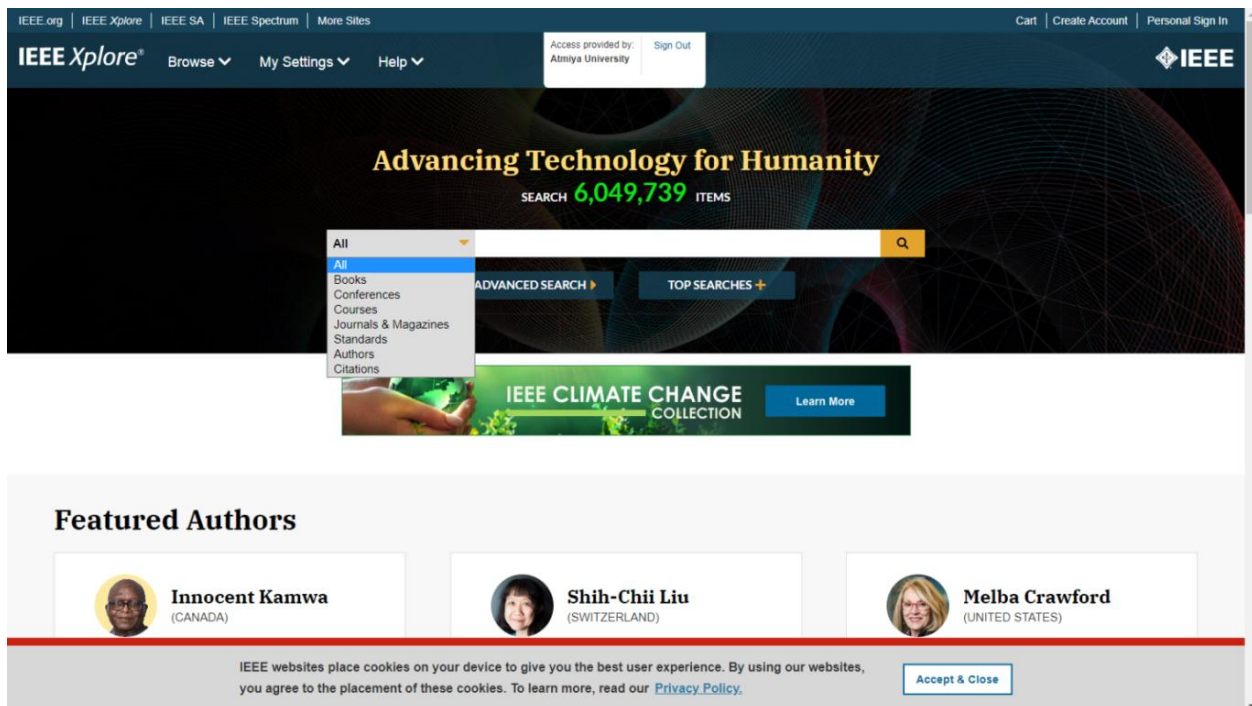
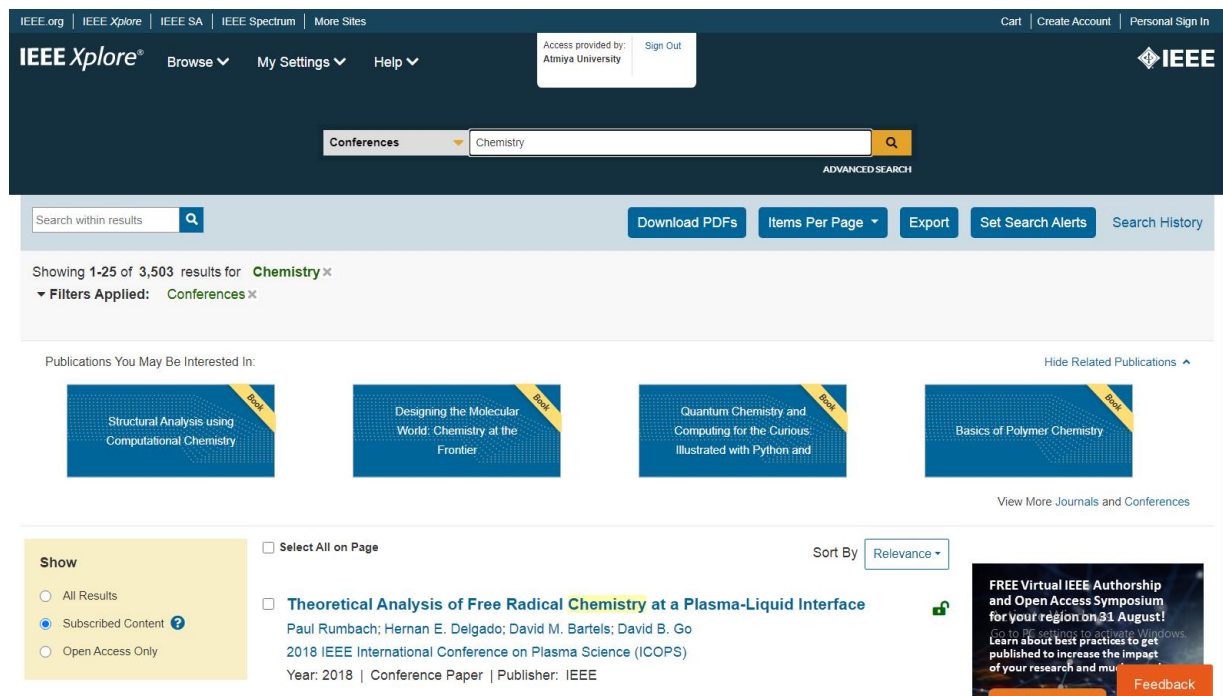
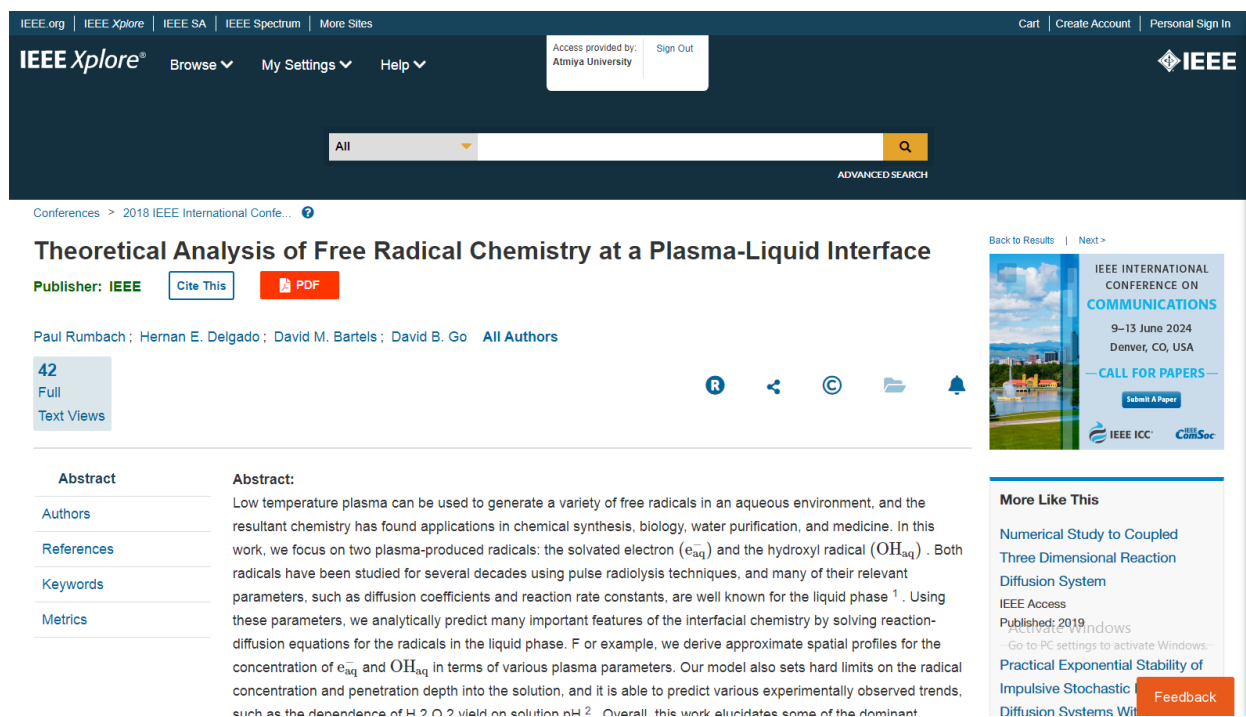


Figure 2: Chemistry subject conferences paper search in subscribe content IEEE Xplore online resource



The screenshot shows the IEEE Xplore search interface. The search bar contains 'Chemistry' and the category is set to 'Conferences'. The results show 1-25 of 3,503 results. A filter for 'Conferences' is applied. Below the search results, there are four book covers: 'Structural Analysis using Computational Chemistry', 'Designing the Molecular World: Chemistry at the Frontier', 'Quantum Chemistry and Computing for the Curious: Illustrated with Python and', and 'Basics of Polymer Chemistry'. A sidebar on the left shows 'Show' options: 'All Results', 'Subscribed Content' (selected), and 'Open Access Only'. The main result displayed is 'Theoretical Analysis of Free Radical Chemistry at a Plasma-Liquid Interface' by Paul Rumbach, Hernan E. Delgado, David M. Bartels, and David B. Go, published in 2018 at the IEEE International Conference on Plasma Science (ICOPS).

Figure 3: Chemistry subject conferences paper abstract in subscribe content IEEE Xplore online resource



The screenshot shows the abstract page for the paper 'Theoretical Analysis of Free Radical Chemistry at a Plasma-Liquid Interface'. The publisher is IEEE. The authors listed are Paul Rumbach, Hernan E. Delgado, David M. Bartels, and David B. Go. The abstract text reads: 'Low temperature plasma can be used to generate a variety of free radicals in an aqueous environment, and the resultant chemistry has found applications in chemical synthesis, biology, water purification, and medicine. In this work, we focus on two plasma-produced radicals: the solvated electron (e_{aq}^-) and the hydroxyl radical (OH_{aq}^\cdot). Both radicals have been studied for several decades using pulse radiolysis techniques, and many of their relevant parameters, such as diffusion coefficients and reaction rate constants, are well known for the liquid phase¹. Using these parameters, we analytically predict many important features of the interfacial chemistry by solving reaction-diffusion equations for the radicals in the liquid phase. For example, we derive approximate spatial profiles for the concentration of e_{aq}^- and OH_{aq}^\cdot in terms of various plasma parameters. Our model also sets hard limits on the radical concentration and penetration depth into the solution, and it is able to predict various experimentally observed trends, such as the dependence of H_2O_2 yield on solution pH². Overall, this work elucidates some of the dominant



Figure 4: Chemistry subject conferences paper full text in subscribe content IEEE Xplore online resource

